## POLYMETHINE DYE DERIVATIVES OF PYRAZOLO[4,5-d]THIAZOLE

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Khimiya Geterotsiklicheskikh Soedinenii, Vol. 3, No. 3, pp. 492-496, 1967

UDC 547.78+668.8

Quino(2)monomethine-, carbo-, mero-, dimero-, and rhodacyanines, as well as styryl dyes containing 4-methyl-6-phenyl- and 4,6-diphenl-pyrazolo[4,5-d]thiazoles groupings are prepared. Study of the colors of the synthesized dyes, showed that dyes containing a pyrazolo[4,5-d]moiety to be considerably more deeply colored, than the corresponding benzothiazole derivatives.

It was previously [1] shown that treatment of 2,4-dimethyl-6-phenyl- and 2-methyl-4,6-diphenylpyrazolo[4,5-d]thiazoles with alkylating agents gives quaternary salts of potential interest as starting materials for preparing various polymethine dyes.

To synthesize a symmetrical carbocyanine derived from 4-methyl-6-phenylpyrazolo[4,5-d]thiazole, an attempt was made to condense a quaternary salt of the base with ethyl orthoformate in pyridine [2] or nitrobenzene (see e.g. [3]). However, it did not prove possible to obtain that particular dye in that way. It was synthesized by reacting the ethylperchlorate of 2,4-dimethyl-6-phenylpyrazolo[4,5-d]thiazole with ethyl orthoformate in acetic anhydride, followed by condensation of the resultant ethylperchlorate of the  $2-\beta$ -ethoxyvinyl derivative (cf. [4]) with a quaternary salt of the 2-methyl substitution product of the same base in acetic anhydride, in the presence of triethylamine. Heating  $2-\beta$ -acetanilinovinylbenzothiazole ethiodide with a quaternary salt of 2,4-dimethyl-6phenyl- or 2-methyl-4, 6-diphenylpyrazolo[4,5-d]thiazole in pyridine-acetic anhydride, or acetic anhydride plus triethylamine (cf. [5]) gave unsymmetrical carbocyanines. The corresponding meso-methyl and -ethyl substituted dyes were obtained by condensing ethylethyl sulfates of the base with chloroethylates of  $2-\beta$ -chloropropenyl- and -butenylbenzothiazoles in dry ethanol in the presence of triethylamine (cf. [6]).

To synthesize a pyrazolothiazolodimethylmerocyanine, a quaternary salt of 2,4-dimethyl-6-phenylor 2-methyl-4,6-diphenylpyrazolo[4,5-d]thiazole was condensed with 3-ethyl-5-acetanilinomethylenerhodanine in dry ethanol in the presence of triethylamine (cf. [7]). It was shown that when the 5-chloromethylene derivative was used in place of the 3-ethyl-5-acetanilinomethylenerhodanine (cf. [8]), the yield of dye was considerably raised. In addition a convenient method of preparing dimethinemerocyanines was reaction of the ethylperchlorate of  $\beta$ -ethoxyvinyl derivatives of the bases with 3-ethylrhodanine, which was effected in acetic anhydride in the presence of triethylamine.

When  $2-\beta$ -anilinovinyl-4-methyl-6-phenylpyrazolo [4,5-d]thiazole methiodide was condensed with 3-ethyl-5-ethylidene- or  $5-\alpha$ -ethoxyenthylidenerhodanine in acetic anhydride plus triethylamine (cf.

[9]), tetramethinemerocyanines were obtained.

Rhodacyanine and dimerocyanine syntheses were effected by heating a quaternary salt of an  $\alpha$ -dimethylmercapto derivative of pyrazolo[4,5-d]-thiazolodimethinemerocyanine with the ethylethyl sulfate of 2-methyl-4,5-diphenylthiazole or 3-butylrhodanine in pyridine (cf., e.g., [10]), respectively.

Styryl and quino(2)monomethinecyanine dyes were obtained respectively by reacting a quaternary salt of 2,4-dimethyl-6-phenylpyrazolo[4,5-d]thiazole with p-dimethylaminobenzaldehyde or 2-ethylmercaptoquinoline ethiodide, in acetic anhydride or dry ethanol and triethylamine.

Tables 1 and 2 give the positions of the absorption maxima of carbocyanine and merocyanine dyes, and for comparison those of the corresponding benzothiazole derivative dyes (measurements were made with a SF-2M spectrophotometer).

The data of Tables 1 and 2 show that on passing from a dye with a benzothiazole group, to the corresponding pyrazolo[4,5-d]thiazolo compound the absorption maximum is displaced towards the longwave region of the spectrum. The hypsochromic shift of the absorption maximum on introducing a methyl or ethyl group at the meso position in the polymethine chain of pyrazolo[4,5-d]thiazolothiacarbocyanine, is of the same order, as for thiacarbocyanine (Table 1).

The hypsochromic shift of the absorption maxima of 4-methyl-6-phenylpyrazolo[4,5-d]thiazolodimethinemerocyanine, as calculated from the absorption maxima of the corresponding oxanine and carbocyanine dyes, (Table 2), is almost the same as the shift for the corresponding dye containing a benzothiazole group, indicating that the basicities of the 4-methyl-6-phenylpyrazolo[4,5-d]thiazole and benzothiazole groups are about the same. Actually the value of A (the concentration of hydrochloric acid, mole/l, bringing about shift, in  $1 \cdot 10^{-5}$  M solution in 57% ethanol, to the diacid salt) found for the diethylate of the symmetrical carbocyanine, a derivative of 4-methyl-6phenylpyrazolo[4,5-d]-thiazole, was 2.2 compared with 2.8 for the corresponding dye containing a benzothiazole group instead.

## EXPERIMENTAL

**2-β-anilinovinyl-4-methyl-6-phenylpyrazolo[4,5-d]**-thiazole ethiodide (I). 1 g (0.003 mole) 2, 4-dimethyl-6-phenylpyrazolo[4,5-d]thiazole ethiodide, 0.4 g (0.003 mole) orthoformic ester, and 0.24 g (0.003 mole) aniline were heated together for 1 hr at 100° C. The resultant crystalline mass was heated with 1 ml dry acetone, the solid filtered off, and washed with

Table 1. Absorption Maxima in Ethanol (in  $m\mu$ ) of Dyes -cH = c - cH = c

7	A		
	Н	CH <sub>3</sub>	C <sub>2</sub> H <sub>5</sub>
4-Methyl-6-phenylpyrazolo[4, 5-d]thiazole* Benzothiazole 4-Methyl-6-phenylpyrazolo[4, 5-d]thiazole 4, 6-Diphenylpyrazolo-[4, 5-d]thiazole	558 558[11] 569 573	543 [12] 556 558	— 547[13] 561

<sup>\*</sup>Z') 4-methyl-6-phenylpyrazolo[4,5-d]thiazole, in the other examples it was benzothiazole.

Table 2. Absorption Maxima and Hypsochromic Shifts of the Dyes, in Ethanol  $C_2H_5$   $C_2H_5$ 

Z	n	A	λ <sub>max</sub> , mμ	Δλ, mμ
4-Methyl-6-phenylpyrazolo[4, 5-d]thiazole Benzothiazole 4, 6-Diphenylpyrazolo[4, 5-d]thiazole 4-Methyl-6-phenylpyrazolo[4, 5-d]thiazole 4-Methyl-6-phenylpyrazolo[4, 5-d]thiazole Benzothiazole	0 0 0 1 1	H H H OC <sub>2</sub> H <sub>5</sub>	538 524[14] 538 624 623 607[15]	25,0 26 [14]

Table 3. Pyrazolo[4,5-d]thiazole Derivative Dyes

	Table 5. Fyrazolo(4, 5-d)mazole Derivative Dyes	oie Derivative	Dycs				
					z.	%	
Dye number	Name	Appearance	Mp (decomp), ° C	Formula	Found	Found Calculated	Yield, %
††  Possi  Dose	1, 3-Diethyl-4-methyl-6-phenylpyrazolo[4 <sub>1</sub> , 5 <sub>1</sub> -d]thiazolo-2'-qui-	Red	245—246	$C_{25}H_{25}IN_4S$	10.06	10.37	5.6
111	3, 3'-Different 1-4'-dimethyl-6, 6'-diphenylpyrazolo[4', 5 <sub>1</sub> -d, 4', 5 <sub>1</sub> -d,	Green	300-301	$C_{29}H_{29}CIN_6O_4S_2$	13.45	13.44	32.2
IV	3, 3'-Dicthyl-4-methyl-6-phenylpyrazolo[4'1, 5'1-d]thiazolothiacar-	Green	235236	$C_{25}H_{25}IN_4S_2$	9.63	67.6	22.8
>	3, 3'-parting round. 3, 3'-parting 4, 8-dinnethyl-6-phenylpyrazolo[4,, 5 <sub>1</sub> -d]thiazolothia-	Bluish-green	250252	$C_{26}H_{27}IN_4S_2$	9.17	9.48	33.9
VI	3, 3', 8-Triethly-4-methyl-6-phenylpyrazolo[4, 5-d]thiazolothiacar-	Dark-green	218219	C <sub>27</sub> H <sub>29</sub> IN <sub>4</sub> S <sub>2</sub>	9.18	9.33	26.7
VII	3, 3'-polyanue roune 3, 3'-polyanue de diphenylpyrazolo[4 <sub>1</sub> , 5 <sub>1</sub> -d]thiazolothiacarbocya-	Green	208209	C <sub>30</sub> H <sub>27</sub> IN <sub>4</sub> S <sub>2</sub>	8.58 6.58	8.82	10.8
VIII	3, 3'-bithyl-4, 6-diphenyl-8-methylpyrazolo[4 <sub>1</sub> , 5 <sub>1</sub> -d]thiazolothia-	Violet needles	174—175	$C_{31}H_{29}IN_4S_2$	8.00 7.00 7.00 7.00	8.64	14.6
ΙX	carbocyanine rounde 3-Ethyl-5-{3'-ethyl-4'-methyl-6'-phenylpyrazolo[4 <sub>1</sub> , 5 <sub>1</sub> -d]thiazo- linylidene-2'-ethylidene}thiazolidine-2-thione-4-one	Reddish- brown plates	219—220	$C_{20}H_{20}N_4OS_3$	12.82	13.08	41.2
X	3-Ethyl-5-{3'-ethyl-4', 6'-diphenylpyrazolo[4, 5 <sub>1</sub> -d]thiazolinyli-dene-9'-ethylidene}thiazolidine-3-thione-4-one	Green	245—247	C <sub>25</sub> H <sub>22</sub> N <sub>4</sub> OS <sub>3</sub>	11,33	11.43	17.0
IX	3-Ethyl-5-{3'-ethyl-4'-methyl-6'-phenylpyrazolo[4, 51-d]thiazol-	Green	204—205	C <sub>22</sub> H <sub>22</sub> N <sub>4</sub> OS <sub>3</sub>	12.25	12.31	26.1
XII	3-Ethyl-5-{3'-ethyl-4'-methyl-6'-planelylpyrazolo[4 <sub>1</sub> , 5 <sub>1</sub> -d]thiazol-	Green	162—163	$C_{24}H_{26}N_4O_2S_3$	11.12	11.22	24.0
XIII*	3, 3'-Diethyl-4-keto-4', 5'-diphenyl-5-{3"-ethyl-4"-methyl-6"-phenylpyrazolof4', 5'-diphazolinylidene-2"-ethylidene}thiazoli-	Brown needles	243244	$C_{38}H_{36}IN_5S_3$	8.89 9.00	9.07	47.3
*VIX	nothiazolocyanine lodide 2-(3'-Butylthiazolidine-2'-thione-4'-one-5'-ylidene)-3-ethyl-5-3"- ethyl-4"-methyl-6"-phenylpyrazolo[4 <sub>1</sub> , 5 <sub>1</sub> -d]thiazolinylidene-2"-	Green needles	255—256	$C_{27}H_{29}O_2N_5S_4$	11.85	11.99	53.9
*\X	ethylidenej-tinazond-4-one 2-p-Dimethylaminostyryl-4-methyl-6-phenylpyrazolo[4 <sub>1</sub> , 5 <sub>1</sub> -d]thia- zole ethiodide	Violet needles	215—216	$C_{23}H_{21}IN_4S$	10.64	10.93	52.8

 $^*\lambda_{\rm max}$  (in etoh) II: 495 mµ, XIII 606 mµ, XIV 585 mµ, XV 526 mµ.

acetone. Yield 0.47 g (33.5%), mp 207°-208° C, yellow needles (ex EtOH) mp 208°-209° C. Found: N 11.48%. 11.27%. Calculated for  $C_{21}H_{21}IN_4S$ : N 11.48%.

Dyes (Table 3). The monomethinecyanine II was synthesized by heating together at 125° C for 15 min 0.001 mole 2,4-dimethyl-6-phenylpyrazolo[4,5-d]thiazole and 0.001 mole 2,4-dimethyl-6-phenylpyrazolo[4,5-d]thiazole and 0.001 mole diethyl sulfate, and then condensing the resultant salt with 0.001 mole 2-ethylmercaptoquinoline ethiodide in 3 ml EtOH, in the presence of 0.002 mole Et<sub>3</sub>N. The symmetrical cabocyanine III was prepared by heating together at 140°C, for 1 hr, 0.001 mole of the base ethylperchlorate, 0.002 mole orthoformic ester, and 1 ml Ac2O, then condensing the resultant product, without first isolating it, with quaternary salt of the same base in 5 ml Ac<sub>2</sub>O in the presence of 0.004 mole Et<sub>3</sub>N, by heating the compounds together for 5 min on a boiling water-bath. The unsymmetrical carbocyanines IV-VIII were synthesized by heating together for 5-10 min, at 100° C, 0.001 mole 2,4-dimethyl-6-phenylor 2-methyl-4,6-diphenylpyrazolo[4,5-d]thiazole ethylethylsulfate, 0.001 mole  $2-\beta$ -anilinovinylbenzothiazole ethyltosylate or  $2-\beta$ -chloropropenyl- and -butenylbenzothiazole methochloride respectively, in 3 ml EtOH or 2 ml EtOH or 2 ml Ac2O and 0.002 mole Et<sub>3</sub>N. Dimethinemerocyanines IX-X were obtained by condensing 0.001 mole base quaternary salt with 0.001 mole 5-chloromethyl-3-ethylrhodanine or 0.001 mole ethylperchlorate of the  $\beta$ -ethoxyvinyl derivatives of the bases with 0.001 mole 3-ethylrhodanine in 2 ml Ac<sub>2</sub>O and 0.006 mole Et<sub>3</sub>N by heating together on a boiling water-bath for 5-10 min. Tetramethinemerocyanines XI-XII were synthesized by reacting 0.005 mole I with 0.005 mole 3-ethyl-5-ethylidene- or  $5-\alpha$ -ethoxyethylidenerhodanine in 3 ml Ac<sub>2</sub>O with 0.002 mole Et<sub>3</sub>N, at 100° C. Rhodacyanine XIII and dimerocyanine XIV were prepared by heating 0.001 mole dimethinemerocyanine IX with 0.002 mole MeSO, for 15 min at 125° C, then condensing the resultant product with 0.001 mole ethylsulfate of 2-methyl-4,5-diphenylthiazole or of 3-butylrhodanine respectively, in 3 ml pyridine, for 20-45 min at 130° C. The styryl dye XV was synthesized by heating 0.0007 mole 2, 4-dimethyl-6-phenyl-pyrazolo[4,5-d]thiazole with 0.0007 mole p-dimethyl-aminobenzaldehyde in 2 ml Ac<sub>2</sub>O for 40 min at 120° C.

All the dyes were purified by recrystallizing from EtOH.

## REFERENCES

- 1. Z. I. Miroshnichenko and M. A. Al'perovich, KhGS [Chemistry of Heterocyclic Compounds], 254, 1965.
  - 2. F. M. Hamer, J. Chem. Soc., 2796, 1927.
- 3. E. Rosenhaur and F. Barlett, Chem. Ber., 62, 2724, 1929.
- 4. G. Bach, Veröffentlichungen des wissenschaftlichen Photo-Labor., Agfa, 9, 94, 1961.
- 5. N. N. Sveshnikov, I. I. Levkoev, and N. I. Shirokova, author's certificate 136375; Byull. izobr. no. 5, 1961.
- 6. N. N. Sveshnikov, I. I. Levkoev, N. I. Shirokova, and N. A. Damir, DAN 148, 1091, 1961.
- 7. Ilford Ltd., British patent no. 428359; C., II, 4656, 1936.
- 8. H. Behringer, E. Dillinger, H. Suter and K. Kohl, Chem. Ber., 91, 2773, 1958.
  - 9. E. B. Knott, J. Chem. Soc., 1490, 1954.
- 10. Ilford Ltd., British patent no. 489335; French patent no. 832 352; C., II, 4164; British patent no. 509917; C., I, 663, 1940.
- 11. W. König and W. Meier, J. pr. Chem., no. 2, 324, 109, 1925.
  - 12. F. M. Hamer, J. Chem. Soc., 3160, 1928.
- 13. L. G. S. Brooker and F. L. White, J. Am. Chem. Soc., 57, 2480, 1935.
- 14. M. V. Deichmeister, Z. P. Sytnik, and E. B. Lifshits, ZhOKh, 22, 166, 1952.
- 15. M. V. Deichmeister, I. I. Levkoev, and E. B. Lifshits, ZhOKh, 23, 1529, 1953.

20 September 1965

Branch of the All-Union Motion Picture Scientific Research Institute, Shostka